

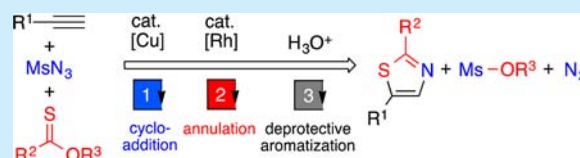
# Facile Synthesis of 2,5-Disubstituted Thiazoles from Terminal Alkynes, Sulfonyl Azides, and Thionoesters

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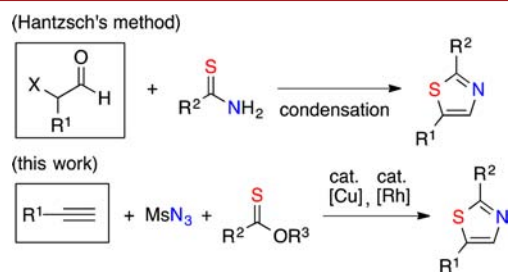
**S** Supporting Information

**ABSTRACT:** A sequential procedure for the synthesis of 2,5-disubstituted thiazoles from terminal alkynes, sulfonyl azides, and thionoesters is reported. A copper(I)-catalyzed 1,3-dipolar cycloaddition of terminal alkynes with sulfonyl azides affords 1-sulfonyl-1,2,3-triazoles, which then react with thionoesters in the presence of a rhodium(II) catalyst. The resulting 3-sulfonyl-4-thiazolines subsequently aromatize into the corresponding 2,5-disubstituted thiazoles by elimination of the sulfonyl group.



A thiazole ring represents a privileged structural motif often found in natural products and pharmaceutically active substances.<sup>1</sup> Moreover, thiazoles multisubstituted with aryl groups are core components constituting materials of interesting optical<sup>2</sup> and electronic properties.<sup>3</sup> The condensation reaction of  $\alpha$ -halocarbonyl compounds with thioamides, named Hantzsch thiazole synthesis, is the most widely used and reliable procedure for their preparation.<sup>4</sup> However, 2,5-disubstituted thiazoles are less accessible with the Hantzsch thiazole synthesis<sup>5</sup> because chemically labile  $\alpha$ -haloaldehydes are required for their synthesis. An alternative method to synthesize those substituted thiazoles is given by palladium-catalyzed C–H arylation reactions of thiazoles with aryl halides, in which an aryl substituent is installed onto a preformed thiazole core.<sup>6</sup> On the other hand, the synthetic methods to directly construct thiazole skeletons possessing 2,5-disubstituents from easily available simple starting substances are still limited.<sup>7,8</sup> We now report a facile method to synthesize 2,5-disubstituted thiazoles from terminal alkynes, sulfonyl azides, and thionoesters (Figure 1). The transformation

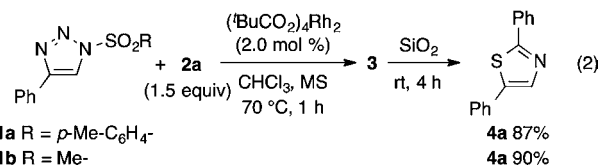
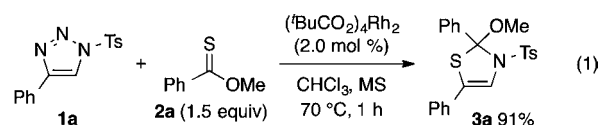
electrophilic carbene carbon and a nucleophilic imino nitrogen in the molecule. They act as the 1,3-dipoles in a formal sense in the reactions with a variety of dipolarophiles, which include alkynes,<sup>11</sup> allenes,<sup>12</sup> nitriles,<sup>13</sup> aldehydes and imines,<sup>14</sup> isocyanates and isothiocyanates,<sup>15</sup> and indoles,<sup>16</sup> affording the corresponding [3 + 2] cycloadducts. In the present study, we examined if thionoesters could serve as the suitable dipolarophiles.<sup>17</sup> First, we prepared *O*-methyl benzothioate (**2a**) from methyl benzoate and the Lawesson's reagent according to a literature procedure.<sup>18</sup> Then, 4-phenyl-1-tosyl-1,2,3-triazole (**1a**, 1.0 equiv) was treated with **2a** (1.5 equiv) in the presence of (<sup>t</sup>BuCO<sub>2</sub>)<sub>4</sub>Rh<sub>2</sub> (2.0 mol %) and 4 Å molecular sieves (MS)<sup>19</sup> in chloroform (2 mL) at 70 °C (eq 1). The triazole **1a** as



**Figure 1.** Construction of 2,5-disubstituted thiazoles.

consists of two catalytic reactions; a copper(I)-catalyzed 1,3-dipolar cycloaddition of terminal alkynes with sulfonyl azides<sup>9</sup> and a rhodium(II)-catalyzed reaction of the resulting 1-sulfonyl-1,2,3-triazoles with thionoesters.

Recently, 1-sulfonyl-1,2,3-triazoles have received much attention as the precursors of  $\alpha$ -imino metal carbene complexes.<sup>10</sup> The generated carbene complex possesses an



consumed in 1 h, and after chromatographic purification using modified silica gel, 2,5-diphenyl-2-methoxy-3-tosyl-4-thiazoline (**3a**) was obtained in 91% isolated yield. The structure of **3a** was confirmed by its single-crystal X-ray analysis. The thiazoline **3a** was labile under acidic conditions to readily aromatize by elimination of a sulfonyl group. Thus, when acidic silica gel was directly added to the reaction mixture containing **3a**, elimination of methyl 4-methylbenzenesulfonate occurred to afford 2,5-diphenylthiazole (**4a**) in 87% yield based on **1a** (eq 2). 1-Mesyl-

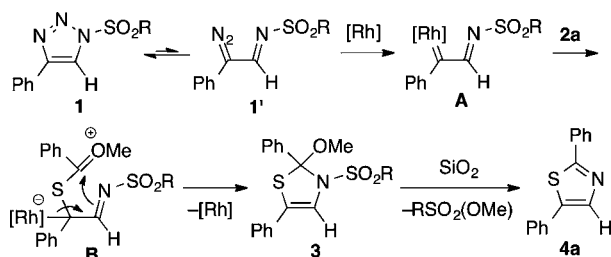
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substituted triazole **1b** also reacted well with **2a** to give a slightly better yield of **4a** (90%).

A plausible mechanism for the production of the thiazole **4a** from 1-sulfonyl-1,2,3-triazole **1** and *O*-methyl benzothioate (**2a**) is depicted in Scheme 1. Initially, a reversible ring–chain

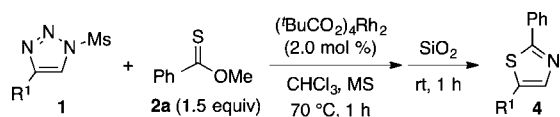
**Scheme 1. Proposed Mechanism for the Formation of Thiazole 4a**



tautomerization of **1** generates  $\alpha$ -diazoimine **1'**, which reacts with rhodium(II) to afford  $\alpha$ -imino rhodium carbene complex **A** with extrusion of molecular nitrogen. The sulfur of **2a** attacks the electrophilic carbene center of **A** to furnish zwitterionic intermediate **B**.<sup>20</sup> The anionic rhodium releases an electron pair, which induces the addition of the imino nitrogen to the carbon of the oxonium ion, forming a five-membered ring. The resulting 4-thiazoline **3** readily aromatizes upon treatment with acidic silica gel by elimination of methyl sulfonate to give the thiazole **4a**.

A variety of triazoles **1** were subjected to the sequential reaction with *O*-methyl benzothioate (**2a**) (Table 1). Triazoles

**Table 1. Rh(II)-Catalyzed Reaction of Various Triazoles **1** with **2a**<sup>a</sup>**



entry	triazole <b>1</b>	product <b>4</b>	yield (%) <sup>b</sup>
	R <sup>1</sup>		
1	<i>p</i> -Me-C <sub>6</sub> H <sub>4</sub> -	<b>1c</b> → <b>4b</b>	96
2	<i>p</i> -MeO-C <sub>6</sub> H <sub>4</sub> -	<b>1d</b> → <b>4c</b>	84
3	<i>p</i> -CF <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> -	<b>1e</b> → <b>4d</b>	90
4	<i>p</i> -Br-C <sub>6</sub> H <sub>4</sub> -	<b>1f</b> → <b>4e</b>	96
5	<i>o</i> -Br-C <sub>6</sub> H <sub>4</sub> -	<b>1g</b> → <b>4f</b>	91
6	<i>p</i> -I-C <sub>6</sub> H <sub>4</sub> -	<b>1h</b> → <b>4g</b>	87
7	<i>p</i> -(pin)B-C <sub>6</sub> H <sub>4</sub> -	<b>1i</b> → <b>4h</b>	67
8	3-Thienyl-	<b>1j</b> → <b>4i</b>	98
9	1-Cyclohexenyl-	<b>1k</b> → <b>4j</b>	87
10	<i>n</i> -Propyl-	<b>1l</b> → <b>4k</b>	49 <sup>c</sup>

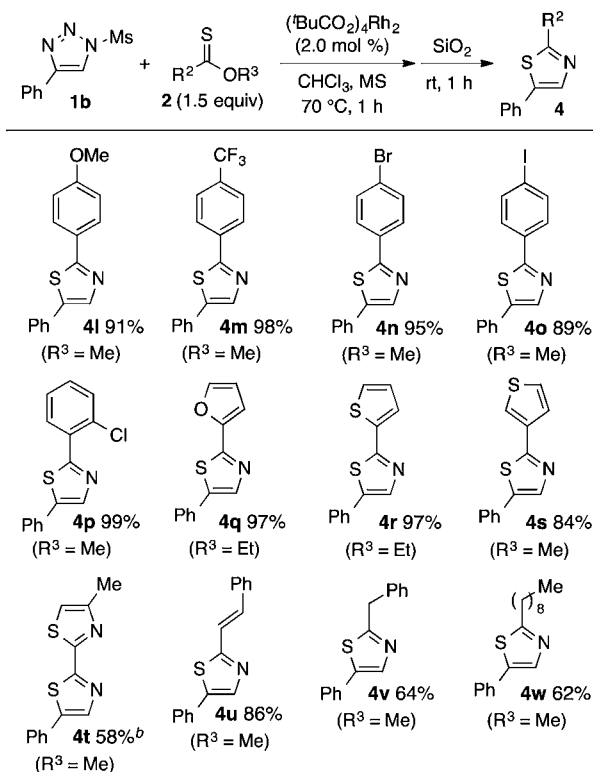
<sup>a</sup>On a 0.20 mmol scale. <sup>b</sup>Isolated yield. <sup>c</sup>Ts-substituted triazole **1l** and **2a** (5.0 equiv) were used.

**1c–i**, possessing an aryl group at the 4-position, afforded the corresponding thiazoles **4b–h** in yields ranging from 67% to 96% (entries 1–7). Notably, the aryl group tolerates a halogen atom and a boron group as the substituents, although the produced halo- and boron-substituted thiazoles are difficult to synthesize with the previously reported palladium-catalyzed C–H arylation reactions of triazoles.<sup>6</sup> Whereas 1-cyclohexenyl-substituted triazole **1k** successfully participated in the sequential reaction

(entry 9), *n*-propyl-substituted triazole **1l** afforded the product **4k** in 49% yield, probably due to a 1,2-hydride shift occurring with the rhodium carbene complex (entry 10).<sup>21</sup>

A diverse array of thionoesters was also readily prepared from the corresponding carboxylic esters by the reaction with the Lawesson's reagent, and they were reacted with the triazole **1b** (Scheme 2). 2-Aryl-5-phenylthiazoles **4l–t** were obtained in

**Scheme 2. Rh(II)-Catalyzed Reaction of **1b** with Various *O*-Alkyl Thioates **2<sup>a</sup>****



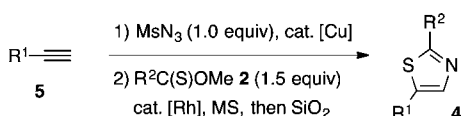
<sup>a</sup>On a 0.20 mmol scale. <sup>b</sup>TsOH·H<sub>2</sub>O was used instead of SiO<sub>2</sub>.

yields ranging from 58% to 99%. Styryl-substituted thionoester was also effectively converted into the product **4u** in 86% yield. In the case of alkyl-substituted thionoesters, the aromatization process from intermediate 4-thiazolines was slightly more sluggish to afford the products **4v** and **4w** in moderate yields.

The synthetic usefulness of the present reaction was demonstrated by its successful integration into a one-pot procedure which directly started from readily available terminal alkynes **5** (Table 2). First, a solution of **5** (1.0 equiv), mesyl azide (1.0 equiv), and CuTC (5.0 mol %) in chloroform was stirred for 8 h at room temperature, generating 1-mesityl-substituted triazoles **1**. Second, thionoesters **2** (1.5 equiv) and (tBuCO<sub>2</sub>)<sub>4</sub>Rh<sub>2</sub> (2.0 mol %) were added to the same vessel, which was then heated at 70 °C for 1 h. Finally, acidic silica gel was added to the reaction mixture to promote deprotective aromatization. The corresponding thiazoles **4** were obtained in overall yields ranging from 80% to 85%. Thus, the rhodium-catalyzed annulation reaction in the second step was barely interrupted by the copper catalyst employed in the first step. This one-pot procedure saves a significant amount of time and solvents for a workup/purification procedure.<sup>22</sup>

The one-pot procedure was applied to steroidal substrate **6a** and  $\delta$ -tocopherol-derived substrate **6b** (eqs 3 and 4). The

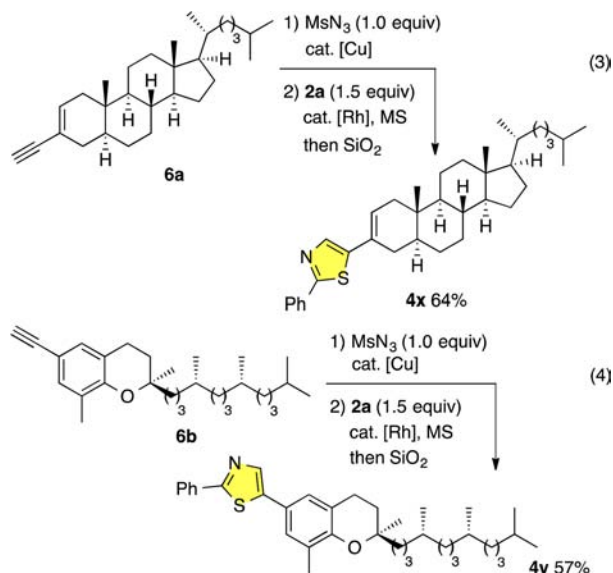
**Table 2. One-Pot Synthesis of Thiazoles 4 Starting from Terminal Alkynes<sup>a</sup>**



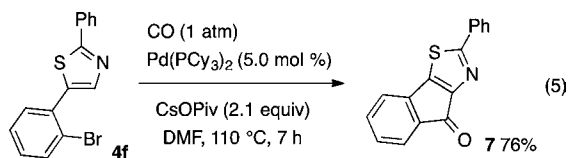
entry	R <sup>1</sup>	5	R <sup>2</sup>	4	yield (%) <sup>b</sup>
1	Ph-	5a	Ph-	4a	80
2	<i>p</i> -MeO-C <sub>6</sub> H <sub>4</sub> -	5b	Ph-	4c	82
3	3-Thienyl-	5c	Ph-	4i	82
4	Ph-	5a	<i>p</i> -MeO-C <sub>6</sub> H <sub>4</sub> -	4l	85
5	Ph-	5a	<i>p</i> -Br-C <sub>6</sub> H <sub>4</sub> -	4n	83

<sup>a</sup>On a 0.20 mmol scale. <sup>b</sup>Isolated yield.

corresponding thiazolyl-substituted derivatives 4x and 4y were successfully obtained in 64% and 57% overall yields based on the starting terminal alkynes 6a and 6b.

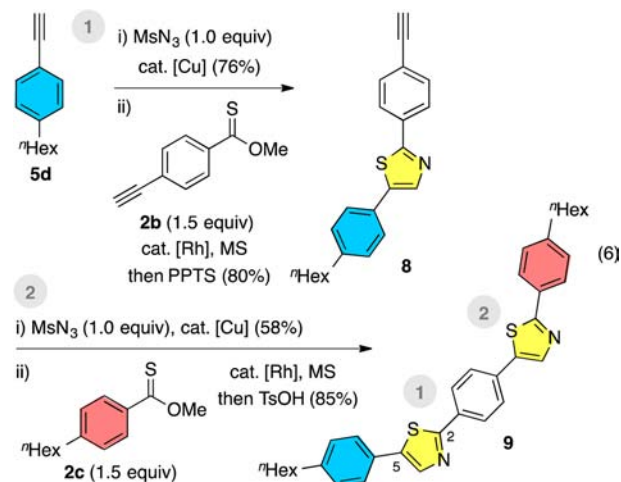


Derivatization of the obtained thiazole was exemplified in eq 5. When the palladium(0)-catalyzed conditions developed by

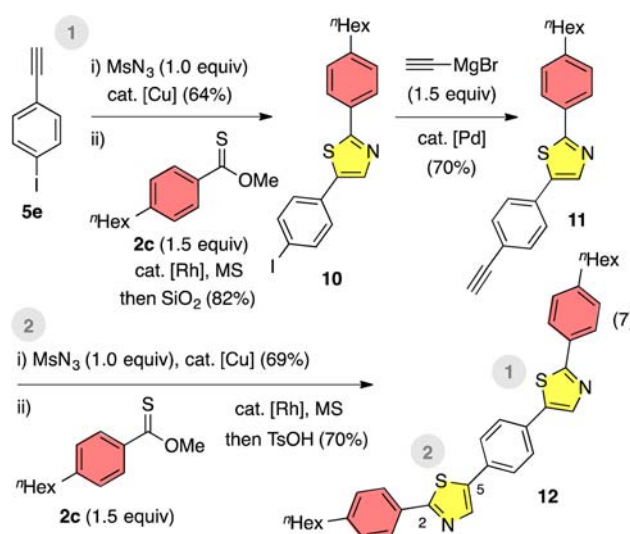


Larock<sup>23</sup> were applied to *o*-bromophenyl-substituted thiazole 4f, the carbonylative cyclization reaction took place to give 4*H*-indeno[2,1-*d*]thiazole-4-one 7.

The terminal alkyne-based thiazole synthesis was further extended to an iterative procedure for the synthesis of linear oligomeric arylene compounds. For example, the ethynyl-substituted benzothioate 2b presents a useful building block for the iterative procedure (eq 6). The first thiazole formation from 5d and 2b was carried out in a stepwise manner to furnish ter(arylene) 8 possessing a terminal ethynyl group. Next, the terminal ethynyl group of the ter(arene) 8 was utilized for the second thiazole formation with 4-hexylbenzothioate 2c to produce the quinque(arylene) 9 consisting of two thiazole and three benzene rings.

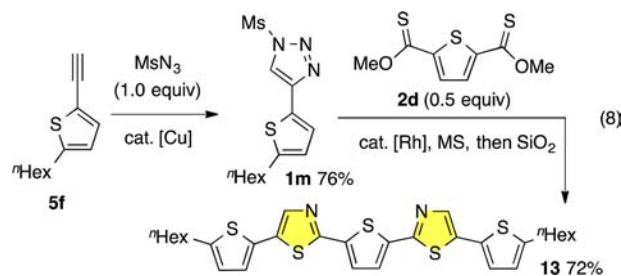


Quinque(arylene) 12 having a different array of two thiazole and three benzene rings was also synthesized (eq 7). Initially, 5-



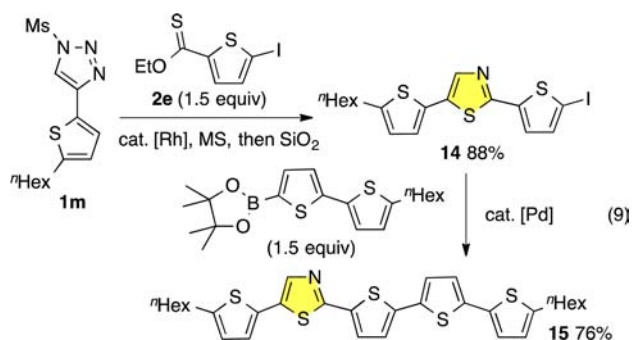
(4-iodophenyl)thiazole 10 was prepared from 5e and 2c, and then a terminal ethynyl group was introduced on the phenyl ring by a palladium-catalyzed coupling reaction,<sup>24</sup> forming 11. The second thiazole formation was carried out using 11 and 2c to produce the quinque(arylene) 12.

Quinque(thiophene/thiazole) oligomers could be efficiently synthesized based on the present thiazole synthesis. The triazole 1m was prepared from 2-ethynylthiophene 5f and mesyl azide. Then, thiophene-2,5-bis(carbomethoxy) 2d (0.1 mmol) was reacted with 1m (0.2 mmol). Double annulation took place to afford symmetrical quinque(thiophene/thiazole) oligomer 13 in 72% yield (eq 8).





Unsymmetrical quinque(thiophene/thiazole) oligomer **15** was constructed from the same triazole **1m** (eq 9). Initially,



the thiazole formation from **1m** and 5-iodothiophene-2-carbonyl thioester **2e** was carried out to furnish iodo-substituted ter(thiophene/thiazole) oligomer **14**. Then, boryl-substituted bi(thiophene) was reacted with **14** in the presence of a palladium catalyst to produce **15** in 76% yield.<sup>25</sup>

In summary, we have demonstrated that thionoesters can act as the dipolarophiles toward  $\alpha$ -imino rhodium(II) carbene complexes and developed a useful method for the synthesis of 2,5-disubstituted thiazoles starting from terminal alkynes. This procedure was successfully applied to late-stage transformation of biorelated derivatives and highly selective synthesis of oligomeric arylene compounds.

## ■ ASSOCIATED CONTENT

### Supporting Information

Experimental procedures, spectral data for the new compounds, and details of the X-ray analysis (CIF). The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b00960.

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### Notes

The authors declare no competing financial interest.

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